

SYNTHESIS OF DERIVATIVES OF NAPHTHO[2,3-b]FURAN-4,9-DIONE

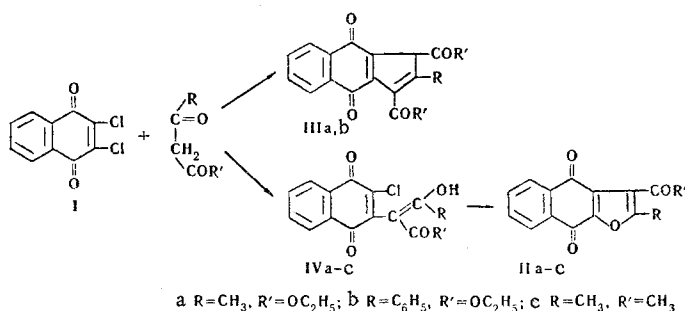
A. N. Grinev and S. A. Zotova

UDC 547.728.2.07

The reaction of 2,3-dichloro-1,4-naphthoquinone with ketoenols has been studied. A series of substituted naphtho[2,3-b]furan-4,9-quinones has been synthesized and some of their reactions have been studied.

We have found that the reaction of 2,3-dichloro-1,4-naphthoquinone (I) with acetoacetic ester forms not 3-ethoxycarbonyl-2-methylnaphtho[2,3-b]furan-4,9-quinone (IIa), as stated earlier [1], but 1,3-diethoxycarbonyl-2-methylnaphtho[2,3-b]cyclopent-1-ene-4,9-quinone (IIIa). Similarly, the condensation of I with benzoylacetic ester leads to 1,3-diethoxycarbonyl-2-phenylnaphtho[2,3-b]cyclopent-1-ene-4,9-quinone (IIIb).

According to the literature [1, 2], IIa can be synthesized by the reaction of sodioacetoacetic ester with I followed by the cyclization of the chloro-1,4-naphthoquinone derivative formed (IVa) by treatment with piperidine. The condensations of I with benzoylacetic ester and acetylacetone were carried out similarly. The cyclization of IVb and IIb was performed by the action of piperidine, and IIc was obtained with a yield of 47% when tetramethylammonium hydroxide was used as catalyst.



With other ketoenols - formylacetone,  $\gamma$ -phenylacetoacetic ester, formylsuccinic ester, and oxaloacetic ester - I forms only resinous products.

We have studied some reactions of the naphtho[2,3-b]furan-4,9-quinone derivatives. The acid V was obtained from 3-ethoxycarbonyl-2-phenylnaphtho[2,3-b]furan-4,9-quinone, and this was converted by decarboxylation into 2-phenylnaphtho[2,3-b]furan-4,9-quinone (VI), which had been synthesized previously by another method [3]. The reductive acetylation of VI led to the diacetyl derivative VII. From the acid chlorides, 2-phenyl- and 2-methyl-4,9-dioxonaphtho[2,3-b]furan-3-carboxamides (VIIIa and b), and also the dimethylamide of 2-methyl-4,9-dioxonaphtho[2,3-b]furan-3-carboxylic acid (VIIIc) were obtained. Under the conditions of reductive acetylation, VIIIb was converted into the 4,9-diacetoxynaphthofuran derivative IX.

The Friedel-Crafts acylation of benzene with 4,9-dioxo-2-phenylnaphtho[2,3-b]furan-3-carbonyl chloride gave 3-benzoyl-2-phenylnaphtho[2,3-b]furan-4,9-quinone (X), obtained previously [2] from I and dibenzoylmethane. The bromination of IIb led to the bromoacetyl derivative XIa or the dibromoacetyl derivative (IXb).

Ordzhonikidze All-Union Scientific-Research Institute of Pharmaceutical Chemistry. Translated from *Khimiya Geterotsiklicheskikh Soedinenii*, Vol. 6, No. 7, pp. 867-870, July, 1970. Original article submitted March 13, 1969.

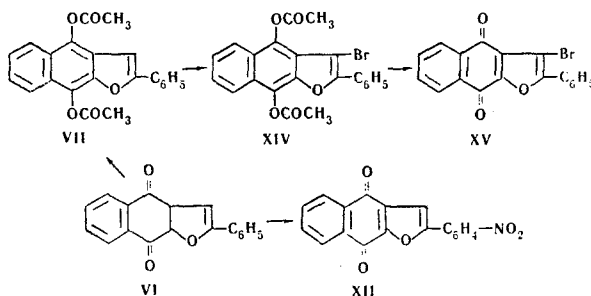
© 1973 Consultants Bureau, a division of Plenum Publishing Corporation, 227 West 17th Street, New York, N. Y. 10011. All rights reserved. This article cannot be reproduced for any purpose whatsoever without permission of the publisher. A copy of this article is available from the publisher for \$15.00.

Some electrophilic substitution reactions have been studied using 2-phenylnaphtho[2,3-b]furan-4,9-quinone (VI) as example.

Compound VI does not undergo Friedel-Crafts acylation when aluminum chloride is used as catalyst and does not undergo bromination; on nitration it forms 2-p-nitrophenylnaphtho[2,3-b]furan-4,9-quinone (XII), the reductive acetylation of which gives 2-p-diacetylaminophenylnaphtho[2,3-b]furan-4,9-quinone (XIII).

It was possible to introduce a bromine atom into the  $\beta$  position of the furan nucleus indirectly. The bromination of 4,9-diacetoxy-2-phenylnaphtho[2,3-b]furan (VII) gave the 3-bromo derivative (XIV), which, after deacetylation and oxidation with oxygen, was converted into 3-bromo-2-phenylnaphtho[2,3-b]furan-4,9-quinone (XV).

The positions of the nitro group and the bromine atom were shown by NMR spectroscopy. The spectrum of VI has a singlet with a chemical shift of 7.04 ppm, which is ascribed to the proton at C<sub>3</sub>. The aromatic protons of the phenyl substituent give a multiplet signal in the 7.6 ppm region. In the spectrum of XV the singlet signal due to the proton at C<sub>3</sub> disappears. The spectrum of XII retains the signal of the proton at C<sub>3</sub> ( $\delta$  7.28 ppm), and the aromatic protons of the phenyl substituent form a quartet with the chemical shifts  $\delta_1$  7.46 ppm and  $\delta_2$  8.13 ppm and the SSCC  $J = 9$  Hz. The shape of the signals shows that the nitro group is present in the para position. The spectra were recorded on a JNM-4H-100 spectrometer with CF<sub>3</sub>COOH as solvent and TMS as standard, using the  $\delta$  scale.



## EXPERIMENTAL

Ethyl  $\alpha$ -(3-chloro-1,4-dioxo-2-naphthyl)acetoacetate (IVa). A solution of 0.76 g (5 mmoles) of the sodium derivative of acetoacetic ester in 5 ml of absolute ethanol was added to a suspension of 1.1 g (0.005 mole) of 2,3-dichloro-1,4-naphthoquinone in 5 ml of absolute ethanol. The mixture was stirred at room temperature for 5 min, and the unchanged I was separated off. The filtrate was acidified with dilute acetic acid, and the precipitate was filtered off. Yield 74%, mp 105–106°C (mp 106–107°C [4]).

Diacetyl(3-chloro-1,4-dioxo-2-naphthyl)methane (IVc) was obtained similarly with a yield of 89%, mp 138–139°C (from 75% acetic acid) (mp 131–132°C [4]). Found %: C 62.04; H 3.94; Cl 12.56%. C<sub>15</sub>H<sub>11</sub>ClO<sub>4</sub>. Calculated %: C 61.97; H 3.81; Cl 12.20.

Ethyl  $\alpha$ -(3-chloro-1,4-dioxo-2-naphthyl)benzoylacetate (IVb) was also obtained similarly with a yield of 70%.\*

3-Ethoxycarbonyl-2-phenylnaphtho[2,3-b]furan-4,9-quinone (IIb). A solution of 0.59 g of IVb in 10 ml of absolute ethanol and 0.6 ml of piperidine was boiled for 1 h and cooled, and the precipitate was filtered off. Yield 94%, mp 182–183°C (from ethanol). Found %: C 72.81; H 4.08. C<sub>21</sub>H<sub>14</sub>O<sub>5</sub>. Calculated %: C 72.83; H 4.07.

3-Acetyl-2-methylnaphtho[2,3-b]furan-4,9-quinone (IIc). A mixture of 0.35 g (5.1 mmoles) of sodium ethoxide in 10 ml of absolute ethanol and 0.93 g (6.1 mmoles) of tetramethylammonium bromide was boiled for 5 min and was then cooled, and 1.45 g (5 mmoles) of IVc was added. The reaction mixture was boiled for 1 h and cooled, and the precipitate was filtered off. Yield 48%, mp 208–209°C (from acetic acid) (mp 202°C [2]). Found %: C 70.68; H 4.28. C<sub>15</sub>H<sub>10</sub>O<sub>4</sub>. Calculated %: C 70.86; H 3.96.

\*The yield is given for the liquid technical product, which was used for further reactions without additional purification.

1,3-Diethoxycarbonyl-2-phenylnaphtho[2,3-b]cyclopent-1-ene-4,9-quinone (IIIb). A mixture of 1.1 g (5 mmoles) of I and 2.13 g (10 mmoles) of the sodium derivative of benzoyl acetic ester in 10 ml of absolute ethanol was boiled for 1 h. After cooling, the precipitate was separated off and crystallized from glacial acetic acid. Yield 78%, mp 152–153°C. Found %: C 72.29; H 4.77.  $C_{25}H_{20}O_6$ . Calculated %: C 72.10; H 4.84.

1,3-Diethoxycarbonyl-2-methylnaphtho[2,3-b]cyclopent-1-ene-4,9-quinone (IIIa) was obtained similarly with a yield of 85%; mp 153°C (mp 152°C [1]).

4,9-Dioxo-2-phenylnaphtho[2,3-b]furan-3-carboxylic acid (V). A solution of 0.5 g (1.5 mmole) of IIb in 4.5 ml of glacial acetic acid and 0.5 ml of concentrated sulfuric acid was boiled for 3 h. The reaction mixture was cooled and the precipitate was filtered off. Yield 65%, mp 214°C (from ethanol). Found %: C 71.80; H 3.40.  $C_{19}H_{10}O_5$ . Calculated %: C 71.70; H 3.17.

2-Phenylnaphtho[2,3-b]furan-4,9-quinone (VI). A mixture of 1 g (3.1 mmole) of V and 10 ml of 85% orthophosphoric acid was heated at 209°C for 2–3 h. The reaction mixture was poured onto ice, and the precipitate was filtered off. Yield 70%, mp 253–254°C (from acetic acid) (mp 246.5–247.5°C [3]). Found %: C 79.06; H 3.83.  $C_{18}H_{10}O_3$ . Calculated %: C 78.82; H 3.67.

4,9-Diacetoxy-2-phenylnaphtho[2,3-b]furan (VII). Over 1 h, 2 g of zinc dust was added to a boiling solution of 0.45 g (1.6 mmole) of VI in a mixture of 20 ml of pyridine and 20 ml of acetic anhydride. The reaction mixture was boiled for 1 h and was poured into 400 ml of cold water. On the following day the precipitate was filtered off. Yield 85%, mp 217–218°C (from acetic acid). Found %: C 73.23; H 4.43.  $C_{22}H_{16}O_5$ . Calculated %: C 73.32; H 4.47.

4,9-Dioxo-2-phenylnaphtho[2,3-b]furan-3-carboxamide (VIIIa) was obtained by the action of aqueous ammonia on the chloride of the acid V with a yield of 75%, mp 304°C (from ethanol). Found %: C 71.57; H 4.80; N 4.51.  $C_{19}H_{11}NO_4$ . Calculated %: C 71.92; H 3.49; N 4.42.

2-Methyl-4,9-dioxonaphtho[2,3-b]furan-3-carboxamide (VIIIb) was obtained similarly. Yield 60%, mp 320°C (from acetic acid). Found %: C 65.73; H 3.88.  $C_{14}H_9NO_4$ . Calculated %: C 65.88; H 3.55.

The dimethylamide of 2-methyl-4,9-dioxonaphtho[2,3-b]furan-3-carboxylic acid (VIIIc) was obtained by the action of an aqueous solution of dimethylamine on the acid chloride with ice cooling. Yield 81%, mp 219°C (from ethanol). Found %: C 68.16; H 4.45; N 5.01.  $C_{16}H_{13}NO_4$ . Calculated %: C 67.84; H 4.63; N 4.94.

Under the conditions of reductive acetylation, VIIIb was converted into 4,9-diacetoxy-2-methylnaphtho[2,3-b]furan-3-carboxamide (IX) in a yield of 88%, mp 228°C (from acetic acid). Found %: C 63.13; H 4.20; N 3.86.  $C_{18}H_{15}NO_6$ . Calculated %: C 63.34; H 4.43; N 4.10.

3-Benzoyl-2-phenylnaphtho[2,3-b]furan-4,9-quinone (X). In portions, 0.75 g of aluminum chloride was added to a solution of 4,9-dioxo-2-phenylnaphtho[2,3-b]furan-3-carbonyl chloride [obtained from 1.59 g (5 mmoles) of V] in 75 ml of dry benzene. The reaction mixture was boiled for 6 h. The complex was decomposed with dilute hydrochloric acid, the layers were separated, and the benzene was distilled off in vacuum. Yield 74%, mp 233–234°C (from ethanol) (mp 231°C [2]).

3-Bromoacetyl-2-methylnaphtho[2,3-b]furan-4,9-quinone (XIa). With stirring, a solution of 0.96 g (6 mmoles) of bromine in 15 ml of dioxane was gradually added to a suspension of 1.27 g (5 mmoles) of IIc in 15 ml of dioxane at 10–15°C. After the end of the addition, the solution was stirred for another 30 min at room temperature. Then the solvent was distilled off in vacuum. Yield 97%, mp 215°C (from acetic acid). Found %: C 53.75; H 2.78; Br 24.29.  $C_{19}H_9BrO_4$ . Calculated %: C 54.08; H 2.72; Br 23.99.

3-Dibromoacetyl-2-methylnaphtho[2,3-b]furan-4,9-quinone (XIb). A suspension of 0.87 g (3.4 mmoles) of IIb in 20 ml of carbon tetrachloride was treated with 1.1 g (6.9 mmoles) of bromine, and the reaction mixture was left at room temperature for two days. Then the precipitate was filtered off, yield 85%, mp 225°C (from acetic acid). Found %: Br 38.40.  $C_{15}H_8Br_2O_2$ . Calculated %: Br 38.78.

2-p-Nitrophenylnaphtho[2,3-b]furan-4,9-quinone (XII). With ice cooling, 0.2 ml of nitric acid (d 1.343) was added to a solution of 0.55 g (2 mmoles) of VI in 4 ml of concentrated sulfuric acid. The reaction mixture was stirred at room temperature for 3 h and poured onto ice, and the precipitate was filtered off. Yield 39%, mp 339°C (from acetic acid). Found %: C 67.43; H 2.82; N 4.35.  $C_{18}H_9NO_5$ . Calculated %: C 67.71; H 2.84; N 4.39.

4,9-Diacetoxy-2-p-diacetylaminophenylnaphtho[2,3-b]furan (XIII) was obtained from XII under the conditions of reductive acetylation. Yield 52%, mp 190-191°C (from ethanol). Found %: C 67.59; H 4.50.  $C_{26}H_{21}NO_7$ . Calculated %: C 67.96; H 4.60.

4,9-Diacetoxy-3-bromo-2-phenylnaphtho[2,3-b]furan (XIV). At room temperature, 0.36 g (2 mmoles) of bromine in 5 ml of glacial acetic acid was slowly added to a suspension of 0.72 g (2 mmoles) of VII in 30 ml of glacial acetic acid. Stirring was continued for another 3 h, and the precipitate was filtered off. Yield 80%, mp 219-220°C (from acetic acid). Found %: C 59.76; H 3.38; Br 18.51.  $C_{22}H_{15}BrO_5$ . Calculated %: C 60.15; H 3.44; Br 18.19.

3-Bromo-2-phenylnaphtho[2,3-b]furan-4,9-quinone (XV). 0.4 g (0.9 mmole) of XIV in 5 ml of absolute ethanol was heated with a catalytic amount of sodium ethoxide for 2-3 min. Then the reaction mixture was cooled, and oxygen was passed through it at room temperature. The precipitate was filtered off. Yield 56%, mp 207-208°C (from acetic acid). Found %: C 61.30; H 2.67; Br 22.80.  $C_{18}H_9BrO_3$ . Calculated %: C 61.22; H 2.57; Br 22.62.

#### LITERATURE CITED

1. G. A. Reynolds, J. A. Van Allan, and R. E. Adel, *J. Org. Chem.*, 30, 3819 (1965).
2. E. F. Pratt and R. G. Rice, *J. Am. Chem. Soc.*, 79, 5489 (1957).
3. S. C. Hooker and A. Steyermark, *J. Am. Chem. Soc.*, 58, 1202 (1936).
4. F. Michel, *Ber.*, 33, 2404 (1900).